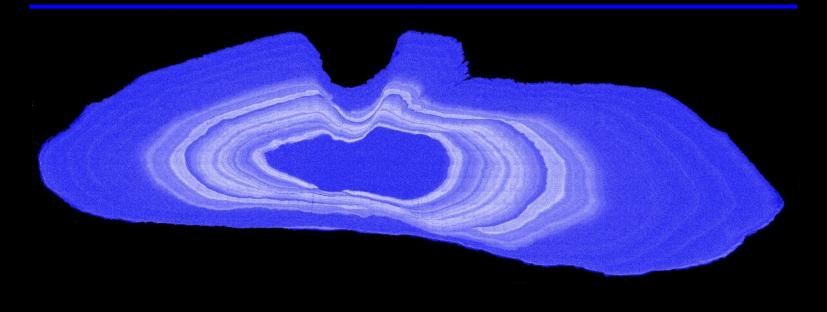
Strontium mark detection and other methods of otolith analysis available at the Advanced Instrumentation Laboratory, University of Alaska Fairbanks.

Ken Severin Advanced Instrumentation Laboratory University of Alaska Fairbanks

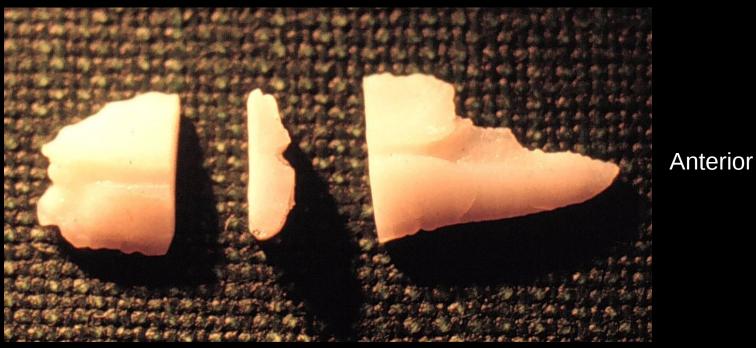




Many thanks to Randy Brown (USFW), Anastasia Ilgen (UAF Chem and BioChem), Dayna Norris (ex-ADFG),Moira Speer, and Karen Spaleta (AIL)

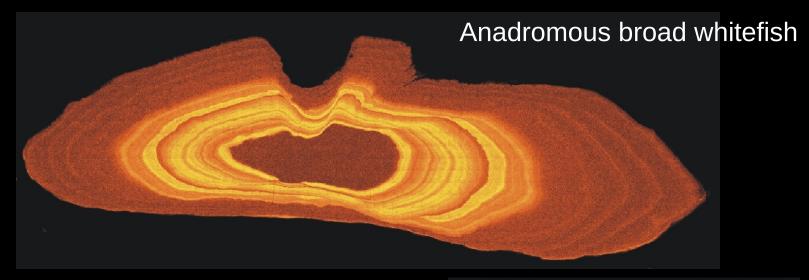
for photographs and help

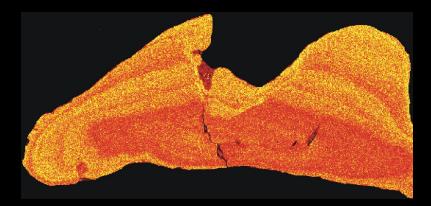




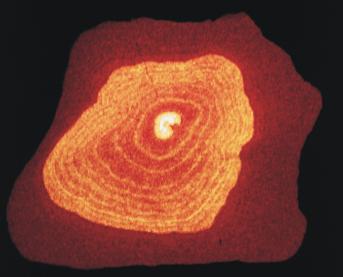
Posterior

Ventral





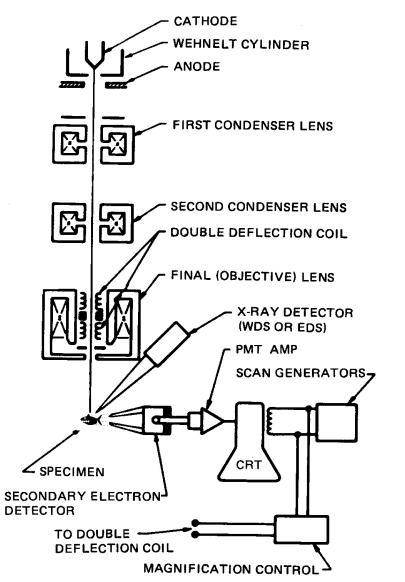
Catadromous Atlantic tarpon

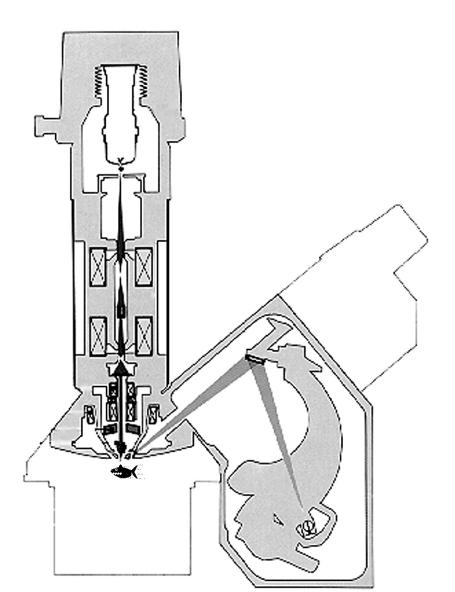


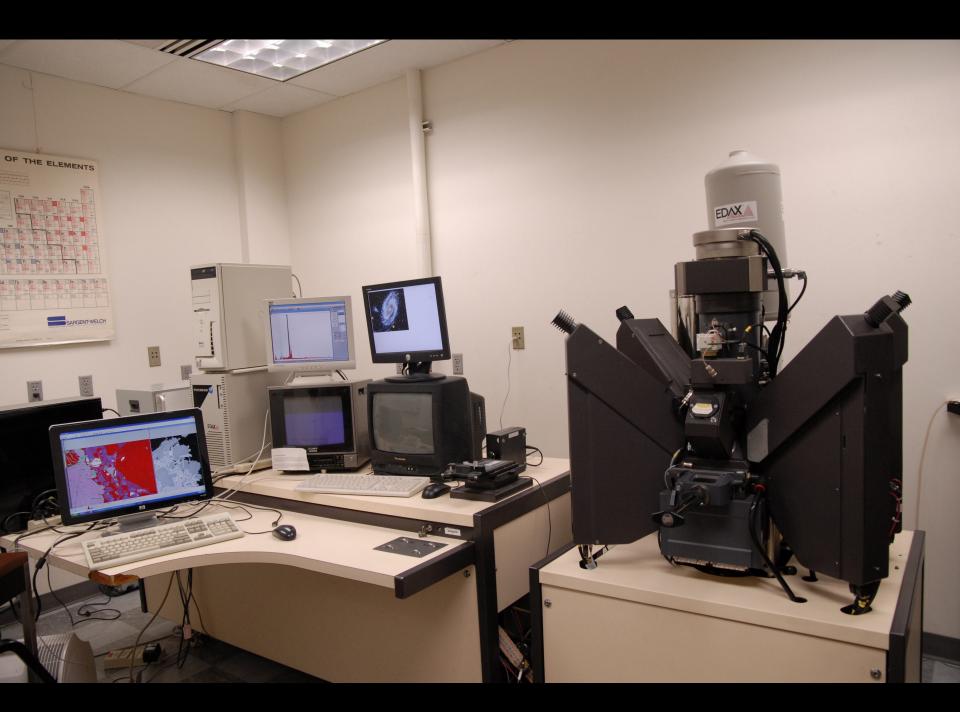
Catadromous European eel

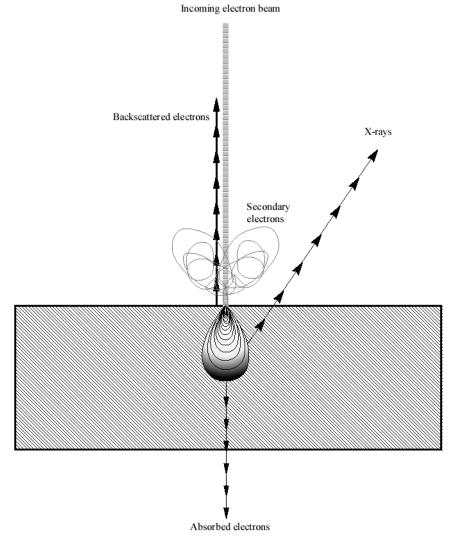
The technology:

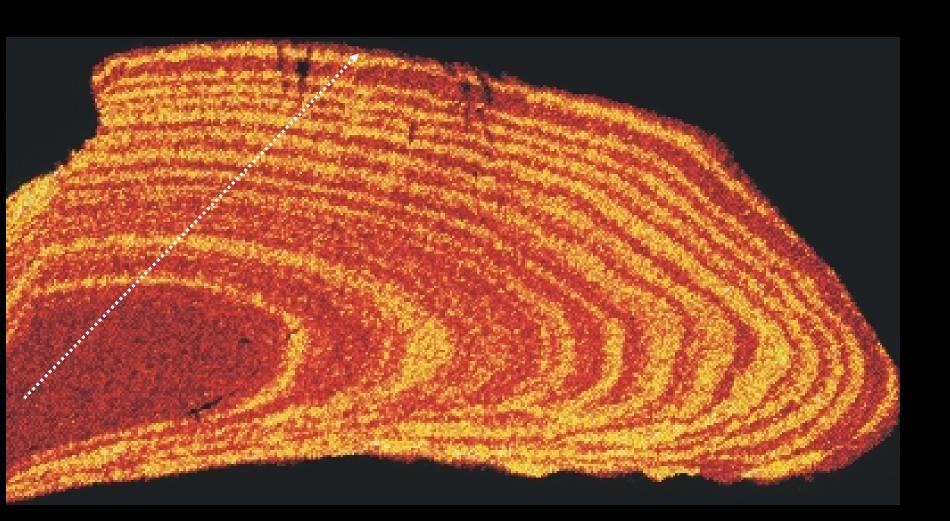
EPMA (Electron Probe Micro Analysis) or EMPA (Electron Micro Probe Analysis)

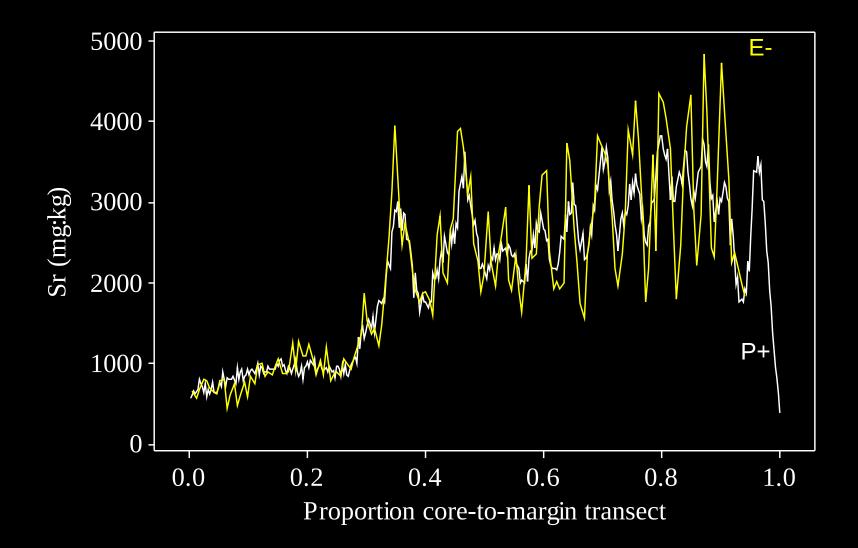


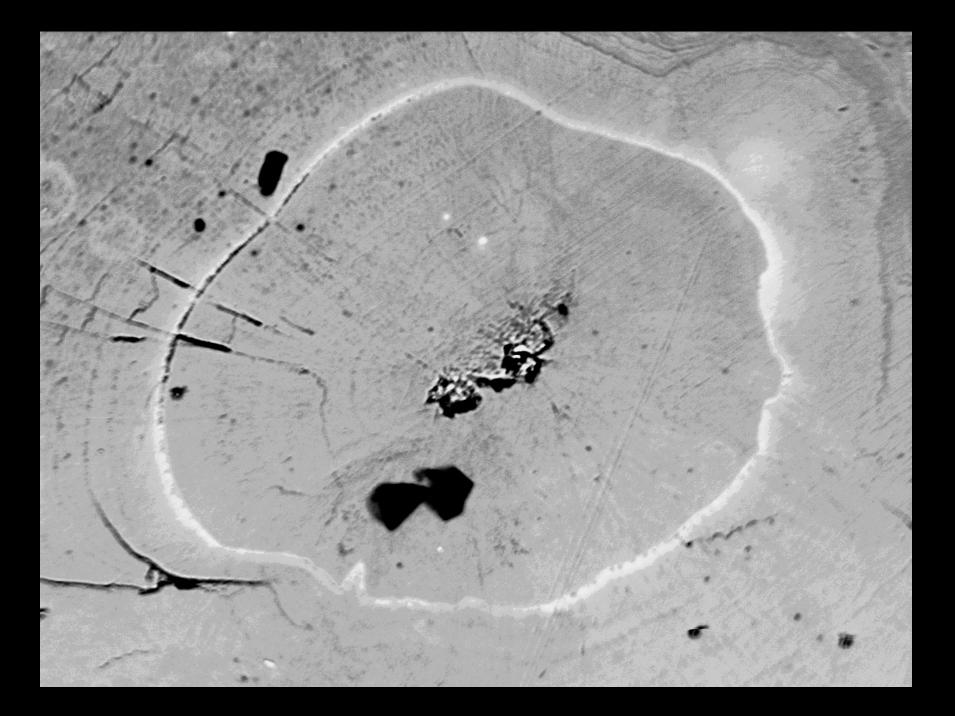












H	Periodic Table of the Elements									© www.elementsdatabase.com				² He			
Li 3	Be	Be ⁴ alkali metals alkali earth metals					 poor metals nonmetals noble gases 					B	C	N	08	F	¹⁰ Ne
11 Na	12 Mg						13 Al	14 Si	15 P	16 S	17 Cl	18 Ar					
19 K	Ca	SC	Ti Ti	V ²³	Cr ²⁴	25 Mn	Fe ²⁶	C0	28 Ni	Cu Cu	Zn Zn	Ga ³¹	Ge ³²	As	³⁴ Se	35 Br	36 Kr
87 Rb	38 Sr	³⁹ Y	⁴⁰ Zr	41 Nb	42 Mo	43 TC	Ru Ru	Rh	46 Pd	Ag	48 Cd	49 In	50 Sn	51 Sb	Te Te	53 	Xe ⁵⁴
Cs	Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	r Ir	Pt	⁷⁹ Au	Hg	81 TI	Pb	⁸³ Bi	⁸⁴ Po	At 85	86 Rn
87 Fr	⁸⁸ Ra	AC	¹⁰⁴ Unq		Unh	¹⁰⁷ Uns	108 Uno	Une									

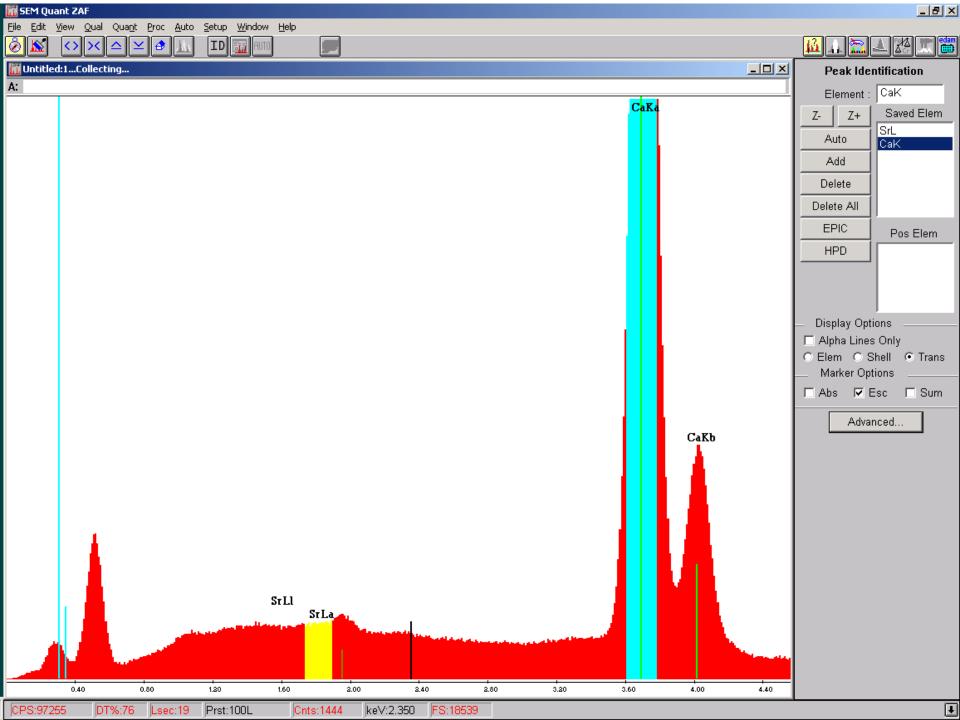
Ce 58	Pr	60 Nd	Pm	82 Sm	Eu	Gd ⁶⁴	Tb ⁶⁵	66 Dy	67 Ho	Er	Tm	Yb	⁷¹ Lu
90	91	92	93	94	95	96	97	Of 98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk		Es	Fm	Md	No	Lr

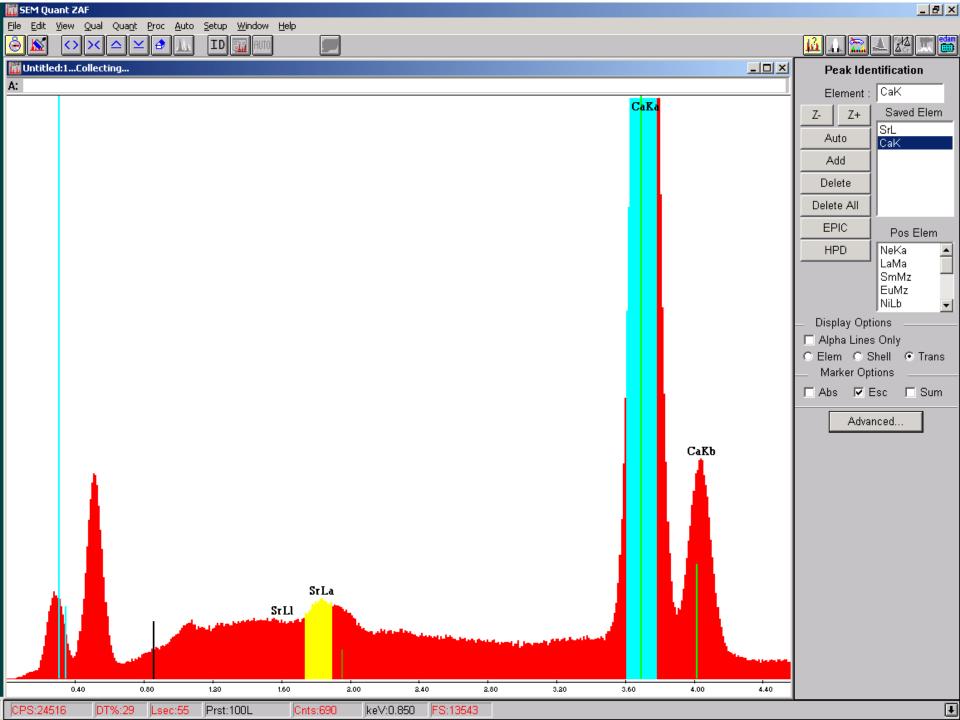
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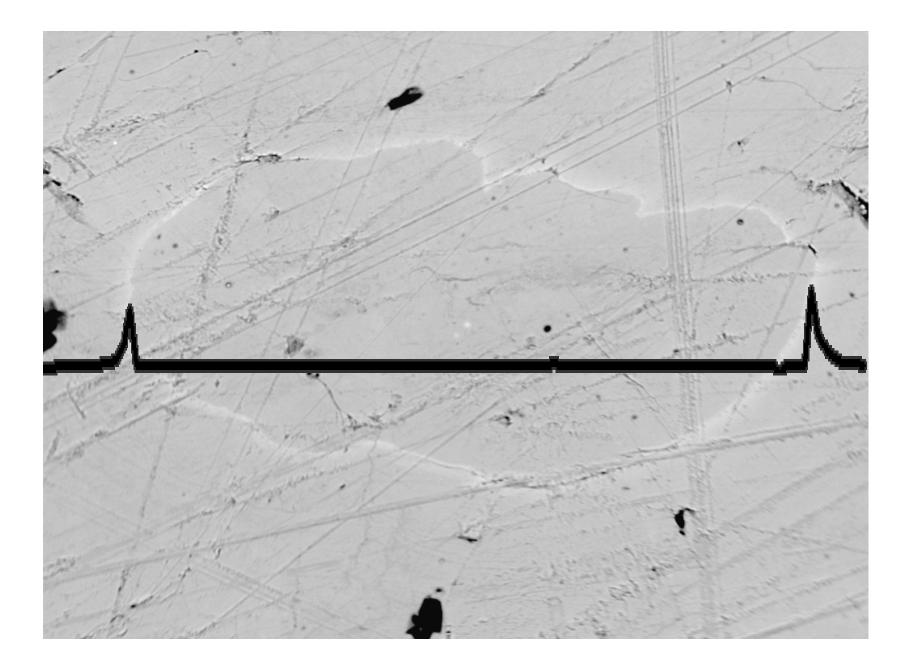












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4												
5	Period:	15										=
6	Date Read:	7/15-16/2009							(mi	n/dd/yyyy))	
7												
8	Otolith#	Marked	Comm	ents from Da	yna No	rris (da	yna	a.norris@alaska.gov) [visi	ual inspecti	on / <mark>micro</mark>	prot	be
9	1	Yes										
10	2	Yes										_
11	3	Yes	[
12	4	Yes										_
13	5	No										_
14	6	Yes										_
15	7	No										_
16	8	No										_
17	9	No										_
18	10	Yes										_
19	11	No						\frown				
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To get to the final product takes:

- Mark Application
- Sample Collection
- Sample Preparation
- Sample Analysis
- Report Preparation
- Report Submission

But I'm not sure that's all

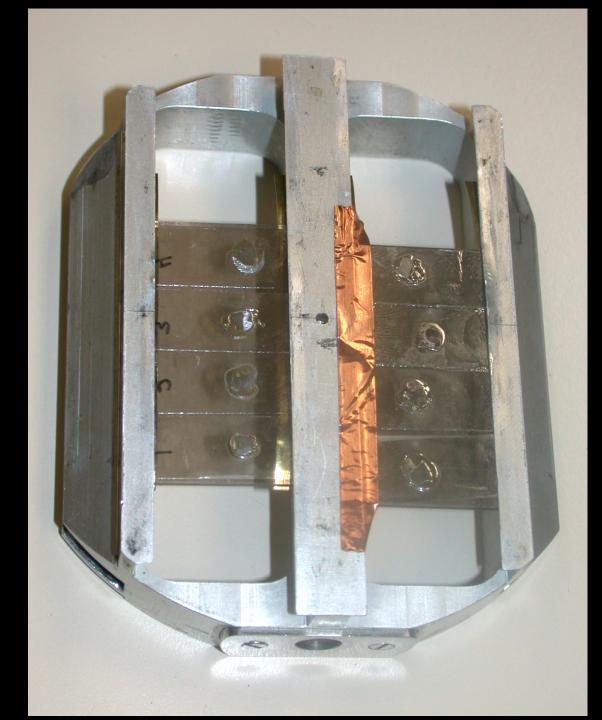
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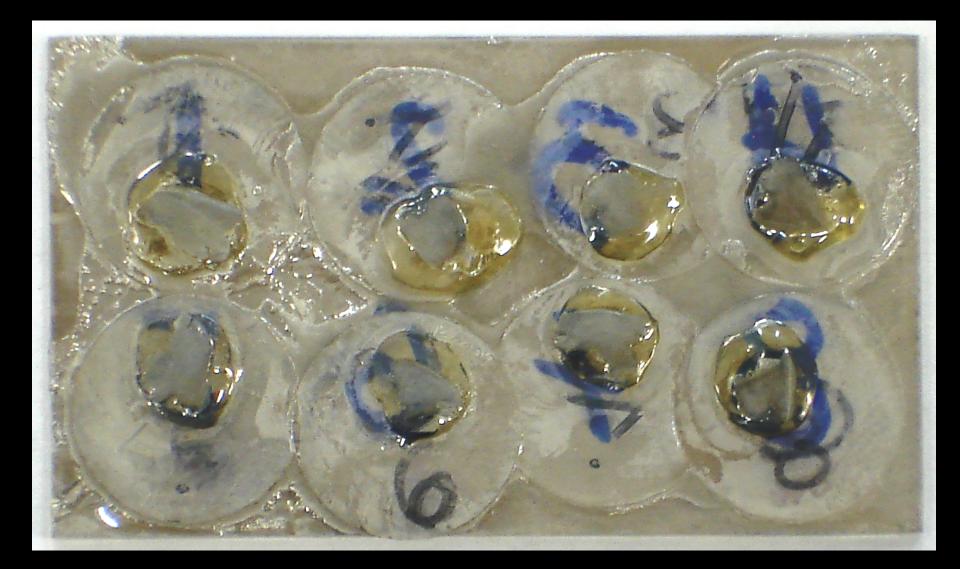
- Mark Application
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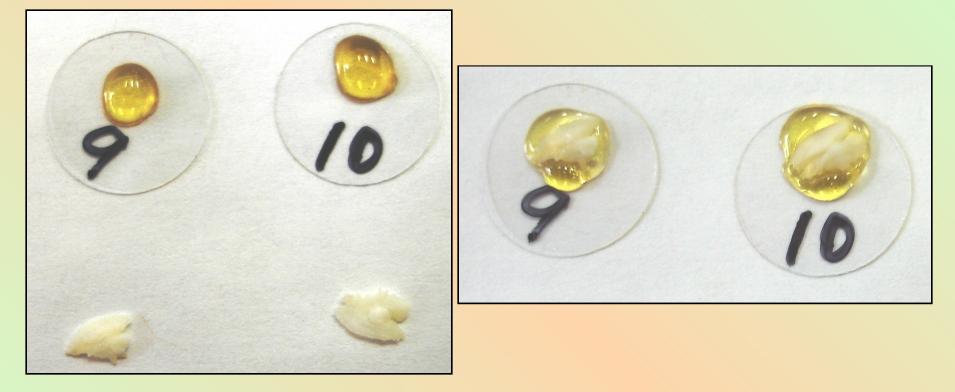
For EM Work the sample should:

- Have the mark exposed
- Be fairly flat and smooth
- (Be electrically conductive)
- Be mounted systematically





Mount otoliths onto cover slips



- Glue should be a clear pool when ready
- Mount sulcus down
- Do not leave on hot plate too long to avoid bubbling

Mount cover slips onto slides



Cover slips are mounted on the slide with wax

Wax is used because:

- it has a lower melting point than the glue
- it holds the cover slip in place while grinding
- comes off easily after cover slip is removed

Grind and polish otoliths



Grinding

- 500-grit SiC paper
- 250 rpm wet grinder

Polishing

- 4000-grit SiC paper
- 250 rpm wet grinder.

Remove cover slips from slides

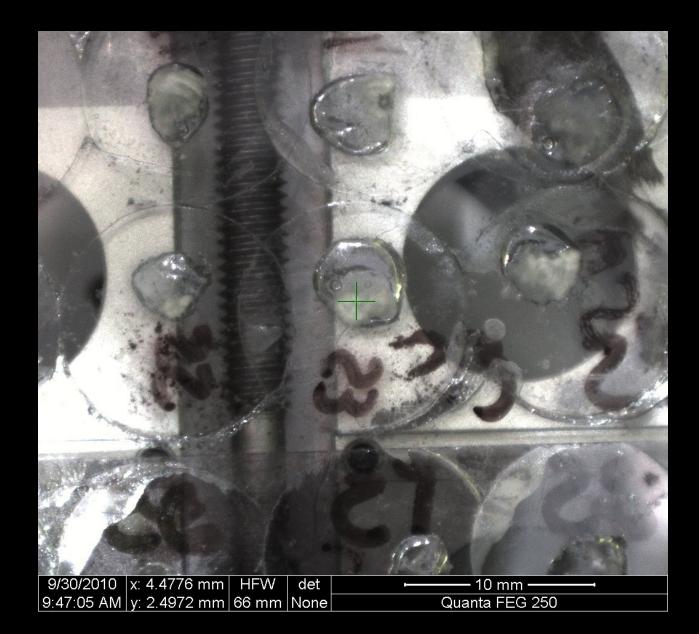


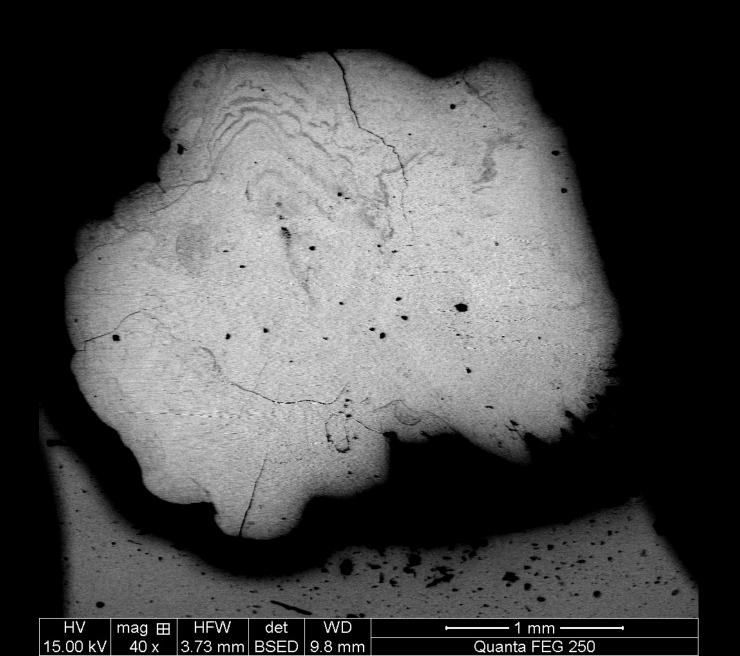
- Melt the wax but not the glue
- Gently push across a paper towel to remove excess wax
- Allow to dry
- DO NOT touch the surface of the otolith

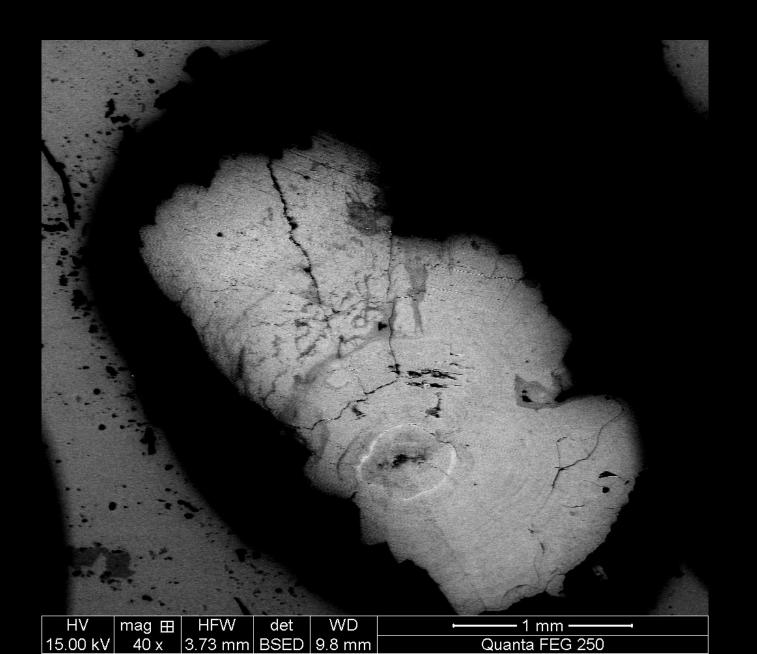
Mount cover slips onto final slide



- DO NOT touch surface
- Blob super glue on slide
- Place cover slip in glue
- Make sure otolith is close to center of slide
- Allow to completely dry-approx. 2 hours







ICP-MS Capabilities and Hyphenated Techniques



Capabilities

- Multi-element detection (metals and non-metals)
- DL down to ppt
- Fast analysis times (full elemental suite in ~4 min)
- Isotopic information

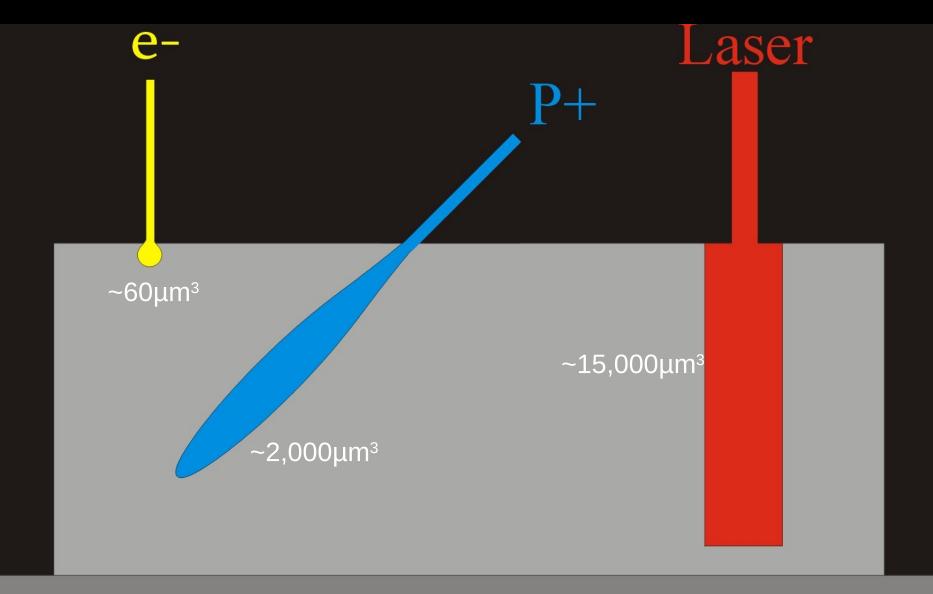
Hyphenated ICP-MS

- LC (liquid chromatography)
- LA (laser ablation)

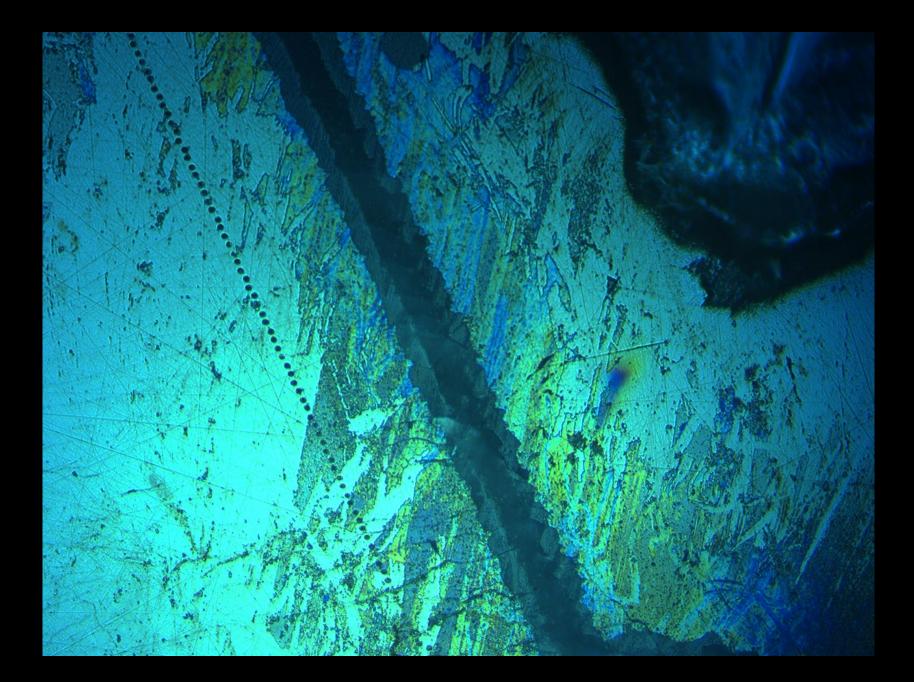


photo by Karen Spaleta

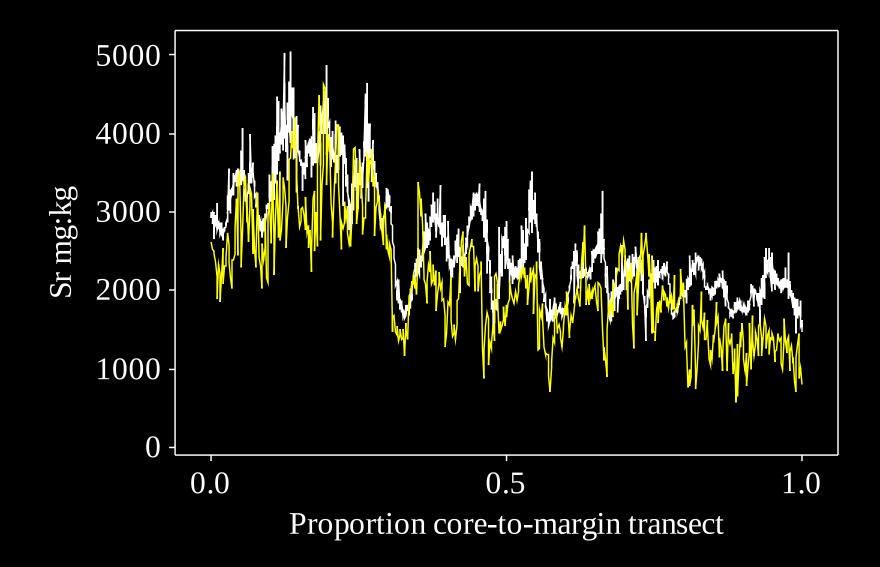
Probe qualities	Electron	Proton	Laser		
Mechanism	ionization	ionization	ablation		
Linear resolution	~5 µm	~5 µm	variable		
Data type	X-ray counts	X-ray counts	lsotope counts		
Count time	25 s/point	3-4 s/point	<1 s/point		
Penetration depth	3 µm	35-50 μm	50-100 μm		
Spatial resolution	high	medium	low		
Surface sensitivity	high	low	low		
Detection level	>100 mg/kg	~1 mg/kg	<1 mg/kg		

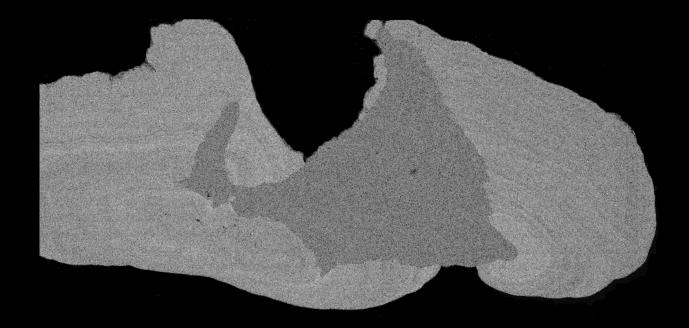


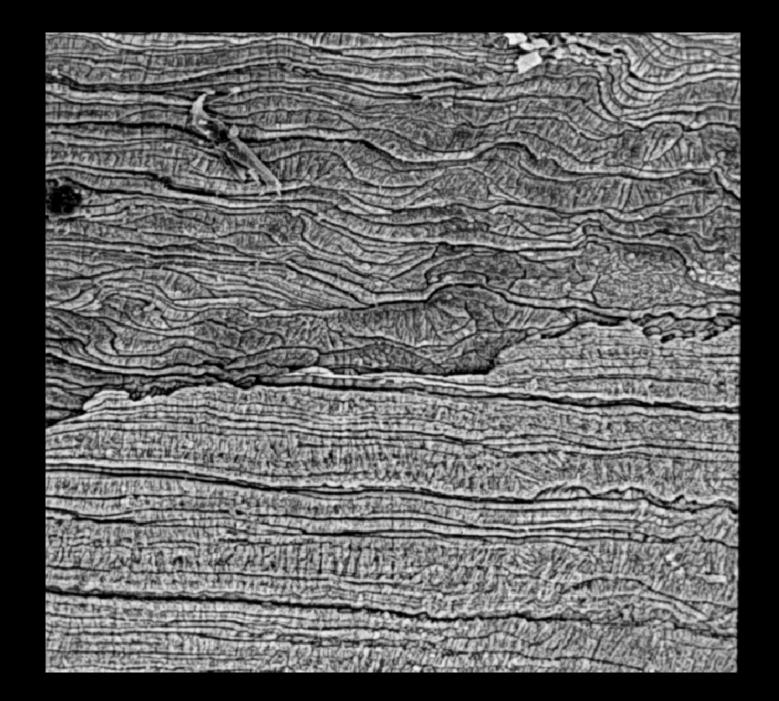
Approximate volumes of material examined per sample point



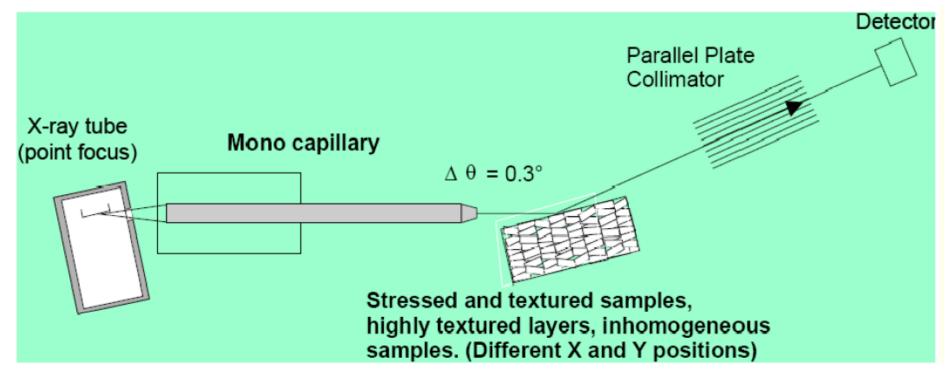
Laser (white) and electron (yellow) Sr distribution data for Barramundi





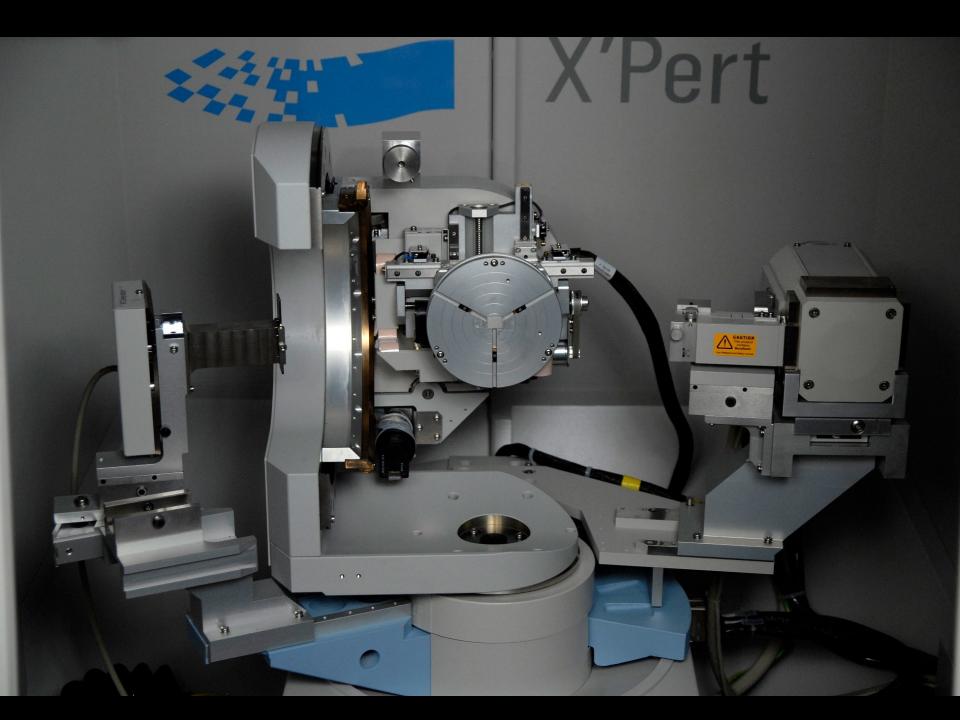


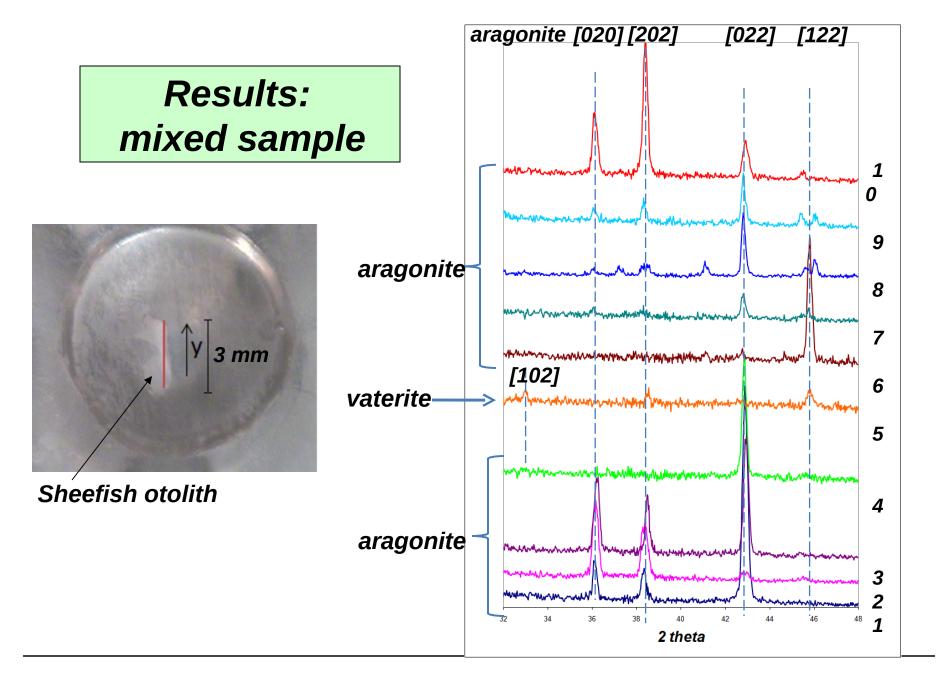
Method: mono-capillary microdiffraction phase analysis

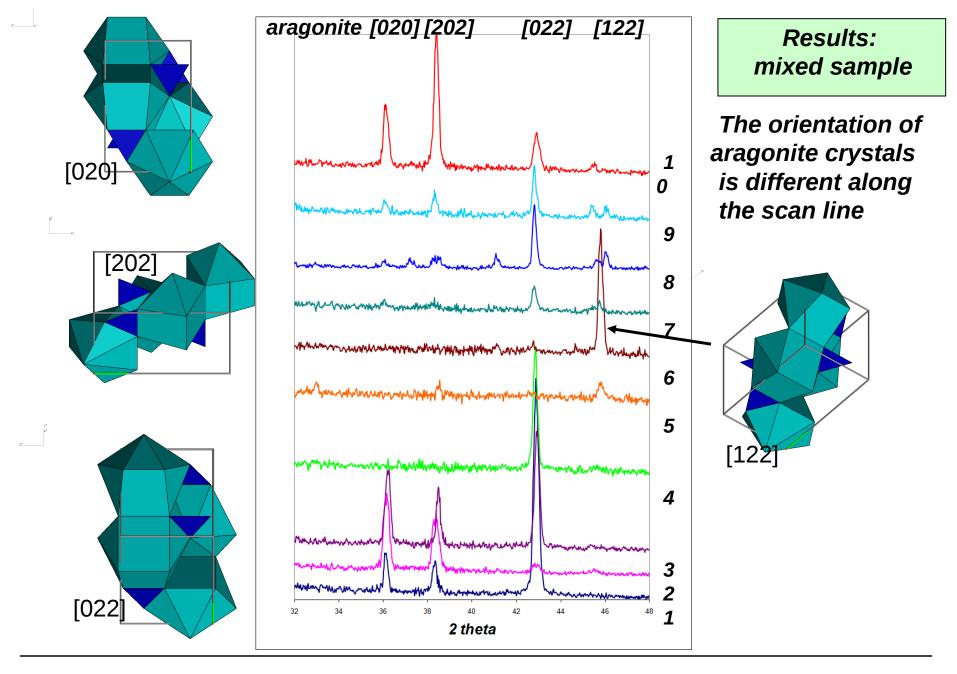


From CHEM 693 Fall 2009 lecture #7 notes

Resolution ~150 μm Challenges: •Alignment of X-ray beam and camera •Finding Z-coordinate so that X-ray beam penetrates sample







X-rays are perpendicular to the page

Some thoughts I've had after the talks on Tuesday

- Assumptions that marked fish act the same as wild fish
- "Compatible with hatchery procedures"
- "Save money in mark assessment"
- Human Error

and probably most important

 What is the question that the interpretation of the mark is supposed to answer, and how good does the answer have to be? www.uaf.edu/ail/

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The Advanced Instrumentation Laboratory at the University of Alaska Fairbanks is a multi-instrument resource for the state of Alaska. It specializes in surface and elemental analysis as well as electron microscopy. In addition to the instrumentation it also houses support sample preparation facilities. AIL is located on the UAF campus in the Reichardt Building.

A to Z

Please forgive us as this page is under construction. You will find several blank pages, and, given our skill with web editing, some things that don't appear quite as nice as we would like.

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Search site/people

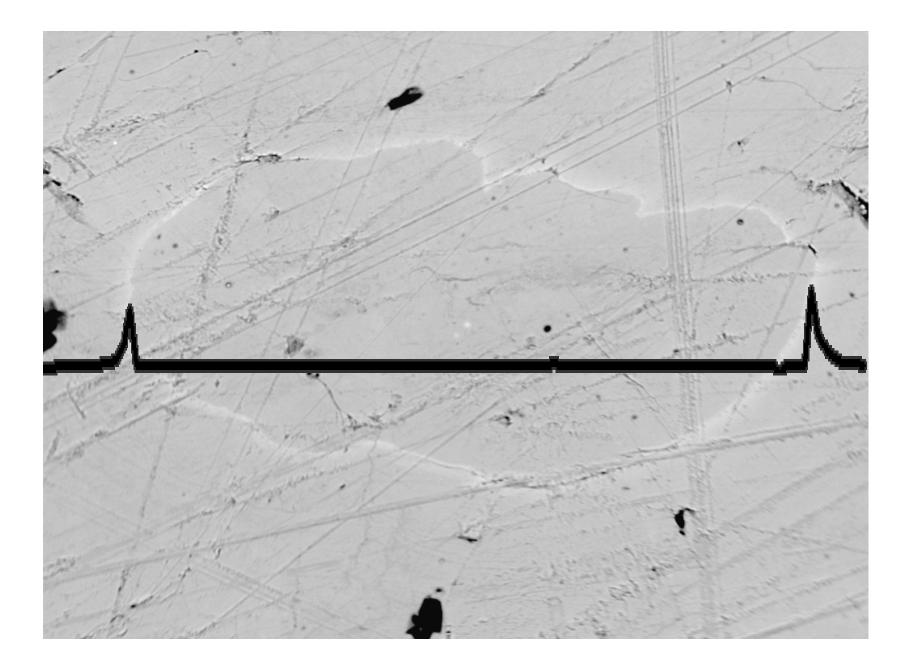
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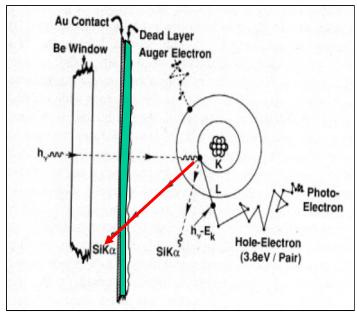
and probably most important

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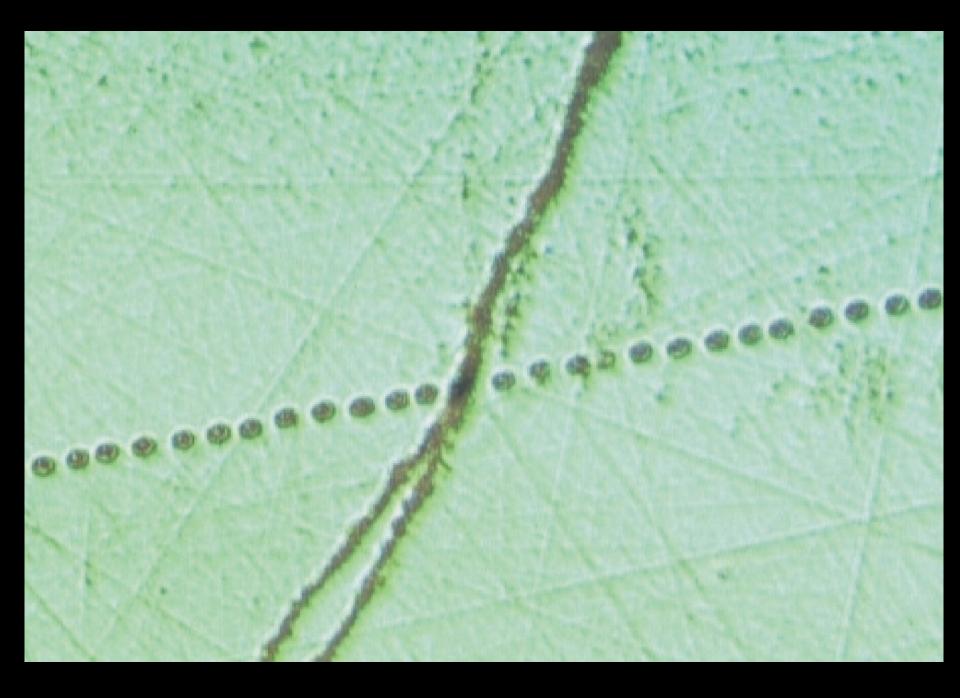


Artifacts: Si-escape peak; Si internal fluorescence peak

There are 2 exceptions to the previous neat explanation of how the Si(Li) detector works.<u>Si-escape peaks</u> are artifacts that occur in a small % of cases, where the Si ka X-ray generated in the capture of the original X-ray escapes out of the detector (red in figure). Since this X-ray removes 1.74 keV of energy, the signal generated (electron-hole pairs) by the incident Xray will be 1.74 keV LOW. This will produce a small peak on the EDS spectrum 1.74 keV **below** the characteristic X-ray peak. Another artifact is the <u>Si internal fluorescence</u> <u>peak</u>, which occurs if an incident X-ray is absorbed in the Si "dead" layer (green



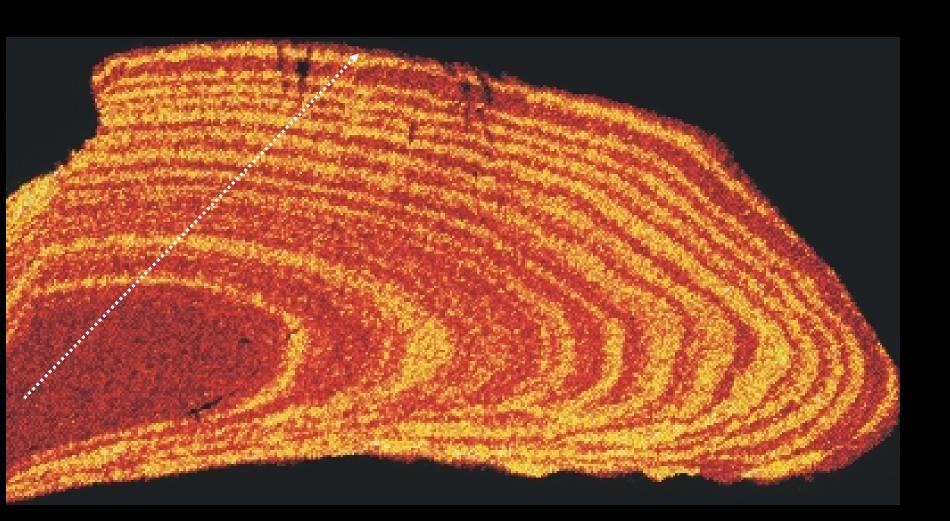
region). This region is "dead" to production of electron-hole pairs, but Si ka X-rays can be produced here which then end up in the "live" part of the detector, and result in a small Si ka EDS peak.



Time Resolution 6 points sample summer material for about 3 weeks per point

Summer 20 weeks 40 µm 2 µm/wk

1 year 50 μm



A Comment:

-Until we understand how and why trace elements are incorporated in otoliths, we will be working with correlations rather than causality. This may not be a problem for marked otoliths where we know what we are putting into otoliths, but will slow the interpretation of natural elemental variability.

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